

Indian Journal of Chemistry Vol. 60B, November 2021, pp. 1522-1527



One-pot synthesis of spiro-3,4-dihydro-2*H*-pyrroles through tandem nucleophilic cyclisation reaction

Nandkishor Chandan

Department of Chemistry, Siddharth College of Arts, Science and Commerce, University of Mumbai, Fort, Mumbai 400 001, India E-mail: nandkishorc@gmail.com

Received 31 March 2020; accepted (revised) 17 September 2021

A very simple and convenient one-pot synthesis of spiro-3,4-dihydro-2*H*-pyrrole has been developed while synthesising the 2,2,5-trisubstituted pyrrolidines. Initially, the Meldrum's acid has been treated with α , β -unsaturated ketones in presence of anhydrous carbonate base and phase transfer catalyst benzyltriethylammonium chloride in acetonitrile to afford the Michael adduct which is readily converted to the corresponding oxime using standard conditions. The crude oxime is treated directly with *p*-toluenesulfonyl chloride and in presence of excess organic base results in tandem nucleophilic cyclisation product spirobicyclic as spiro-3,4-dihydro-2*H*-pyrrole directly, instead the tosyloxime as product.

Keywords: Pyrrole, spiro, oxime, tosyloxime, cyclization, Meldrum's acid, tandem, nucleophilic, α,β-unsaturated ketone

The substituted Pyrroles are the most important and explored heterocyclic compounds as they are found in many natural products and biologically active compounds¹⁻³ as well are privileged scaffolds for pharmaceutical products⁴⁻⁸ since they exhibit a diverse range of biological activities, including analgesic, anti-convulsant. anti-depressant, anti-allergic, anti-diabetic. anti-hyperlipidemic, anti-microbial, anti-fungal, anti-viral, anti-inflammatory, cholesterol-reducing, and anti-tumor properties⁹⁻¹². Therefore, substituted pyrroles are considered as a potential source of biologically active compounds with valuable properties.

The Spiro heterocyclic compounds are considered as a privileged framework because of its rigidity, three-dimensional geometries, and wide distribution in various natural products and synthetic molecules. Presently, these Spiro-heterobicycles compounds are attracting considerable interest in synthetic organic chemistry because of its molecular structure and diverse biological activities¹³. In addition, spiroheterocycle systems impart structural novelty for drug discovery because of its broad range of medicinal properties¹⁴⁻¹⁸, as well important structural motifs that are entrenched in a variety of drugs, including the cardiovascular active Irbesartan^{19,20}, progestogenic agent Drospirenone^{21,22} and potassium-separating diuretic Spironolactone^{23,24}. The normally used approach in the drug designing is to rigidify the ligand conformation by a spiro-ring fusion, which can impact ligand binding entropy upon binding to a protein target. Commonly, bioactive spirocyclic molecules contain five membered ring systems in their structures. In particular, spiro-pyrroles represent important structural motifs that can be found in many biologically active synthetic compounds and natural products^{25,26}. We have recently reported that, under basic conditions, tosyloxime of Michael adduct with diethyl malonate results in the cyclization with C-N bond formation to pyrrolidine embedded with varies functionality^{27,28}. In similar line to introduced the rigidity, we have used Meldrum's acid as source of nucleophile in Michael addition with different α , β-unsaturated ketones which on oximation and followed by tosylation in excess of basic condition results into the Spiro-3,4-dihydro-2H-pyrrole in good yield as one pot synthesis.

Result and Discussion

Spiro-heterobicycles appear in countless natural products and clinically valuable compounds²⁹ possibly among the most challenging structural motifs to synthesise, spirocyclic synthesis has inspired chemists for decades. Meldrum's acid appears to be an attractive reagent in organic synthesis owing to its high acidity, steric rigidity and high reactivity³⁰.

Following the methodology developed²⁷ using oxime ether mediated ring closure, Meldrum's acid could be very useful to synthesise spiro-heterobicyclic rings. Indeed, the spiro-heterobicycle **3** could be helpful to construct non natural amino acid by solvolysis of Meldrum's acid with alcohols would give a diastereomeric mixture of a half acid ester; and there could be further interconverted. The half carboxylic acid can be converted into amine or carbamate, through acid azide formation, and followed by Curtius rearrangement, would give an isocyanate, which could be easily converted into amine or carbamate by hydrolysis or alcohol treatment³¹.

The required spiro-heterobicyclic ring 3 could be synthesised by following the sequence of reactions. The reaction of Meldrum's acid with methyl vinyl ketone in presence of 1.1 eq. of anhydrous K₂CO₃ and eq. of benzyltriethylammonium chloride in 1 acetonitrile as the solvent, at 50-60°C for 8-10 h³² afforded the Michael adduct 1 in 81% yield with some double Michael addition product as by-product which was minimised by using 1.1eq. base. The Michael adduct 1 was readily converted to the corresponding oxime using standard conditions (NH2OH.HCl and Et₃N in EtOH stirred at 50°C for 1 hr) and afforded 2 in 77% yield. The crude oxime 2 was used directly and treatment with p-toluenesulfonyl chloride and base (excess Et₃N in dry CH₂Cl₂ at RT) gave the cyclization product Spiro-3,4-dihydro-2H-pyrrole 3 directly in 78% of yield (Scheme I) instead the expected tosyl oxime (Scheme I).

The resultant Spiro-3,4-dihydro-2*H*-pyrrole **3** is crystalline and this product was fully characterised by single-crystal X-ray crystallographic analysis (Figure 1), clearly demonstrating the fully orthogonal nature of the two heterocyclic rings. The spiroheterobicycle **3** was previously reported by Danheiser³³ and he obtained as oil but NMR data agree.

Similarly the Michael addition reaction with Meldrum's acid and α,β -unsaturated ketones such as 1-phenylbut-2-en-1-one, 1-phenylprop-2-en-1-one, pent-3-en-2-one and chalcone was carried out in presence of anhydrous K₂CO₃ and 1 eq. of benzyltriethylammonium chloride in acetonitrile as the solvent, at 50-60°C for 8-10 hr except chalcone other ketones gave good yield as Michael adduct as shown in Scheme II. The same Michael adduct on oximation using standard condition and followed by treatment with *p*-toluenesulfonyl chloride and base (excess Et₃N in dry CH₂Cl₂ at RT) gave the cyclization product Spiro-3,4-dihydro-2H-pyrrole as 6, 9 and 12 in good yield as shown in Scheme II but as the compounds are solid but the formation of single crystal was so difficult and all compounds are characterised bv Mass and NMR spectra.

Experimental Section

General procedure for the preparation of Michael adduct ⁹⁻¹²

To a stirring solution 7.63 mmol of Meldrumic acid (1.1g) in acetonitrile (20 mL) was added 8.39 mmol



Reagents and conditions: (a) anhydrous K₂CO₃, benzyltriethylammonium chloride, acetonitrile, heated 50-60°C; (b) NH₂OH.HCl, Et₃N, EtOH, heated 50°C; (c) *p*-TsCl, Et₃N, CH₂Cl₂, RT

of anhydrous K_2CO_3 (1.15g) and 7.63 mmol of benzyltriethylammonium chloride (1.73g). The mixture was stirred for 15 min at RT and 7.63 mmol of methyl vinyl ketone (0.64mL) was then added and the resultant mixture was stirred for 8 hr at 50°C. After TLC monitored, the mixture was cooled to RT, the reaction was quenched with water and mixture was washed with diethyl ether and then acidified by 6 N HCl. The adduct was extracted by diethyl ether and dried over Na₂SO₄, to give crude product, which was purified by recrystallization using EtOAc and petroleum ether to afford Michael adduct 1.

General procedure for the preparation of oxime of Michael adduct ^{27,28}

To the solution of 2.33 mmol Michael adduct ketone 1 (500mg) in 10 mL EtOH was added 4.66 mmol of NH₂OH·HCl (183mg) and 4.66 mmol of Et₃N (470mg) and the reaction mixture was heated to 50°C for 1 hr. The progress of the reaction was monitored by TLC and on completion of the reaction mixture was allowed to cool to RT. EtOH was



Figure 1 — X-ray crystal structure for spiro-3,4-dihydro-2H-pyrrole **3**

removed under reduced pressure and the residue was dissolved in EtOAc and H_2O added. The organic layer was separated, washed with brine and dried over Na₂SO₄. Concentration of the organic layer gave the crude product, which was purified by flash column chromatography (eluting with EtOAc : petrol) to afford the oxime product **2**.

General procedure for cyclization to spiro-3,4dihydro-2*H*-pyrrole²⁹

The stirring solution 2.18 mmol of 2 oxime (500mg) in dry 10 mL CH₂Cl₂ at 0°C was treated with 6.54 mmol of Et₃N or pyridine (660mg or 516mg) followed by slow addition 2.61 mmol of ptoluenesulphonyl chloride (494 mg) and reaction was stirred for 1 hrs and it was found in mass spectra showed presence of cyclized product. The reaction mixture was further stirred for 2 hrs and the reaction progress was monitored by TLC and by mass spectra and it was found that instead of tosyloxime the cyclized product was formed and after completion of reaction (1 M) HCl was added and product was extracted by CH₂Cl₂ and the extracts were washed with saturated NaHCO₃ and dried over Na₂SO₄. Concentration of CH₂Cl₂ gave crude product which was purified by flash column chromatography (eluting with EtOAc : petrol) to afford the Spiro-3,4dihydro-2H-pyrrole.

(±) Dimethyl-5-(3-oxobutyl)-1,3-dioxane-4,6-dione, 1^{32}

Following the general procedure 1, to the stirring solution 15.3 mmol of Meldrumic acid (2.21g) in



Reagents and conditions: (a) anhydrous K₂CO₃, benzyltriethylammonium chloride, acetonitrile, heated 50-60°C, (b) NH₂OH.HCl, Et₃N, EtOH, heated 50°C; (c) *p*-TsCl, Et₃N, CH₂Cl₂, RT

acetonitrile (40 mL) was added 16.8 mmol of anhydrous K₂CO₃ (2.31g) and 15.3 mmol of benzyltriethylammonium chloride (3.48g). The mixture was stirred for 15 min at RT and 15.3 mmol of methyl vinyl ketone (1.3mL) was then added and the resultant mixture was stirred for 8 hr at 50°C to give crude product, which was purified by recrystallization using EtOAc and petroleum ether to afford adduct 1 (2.67 g, 81%) as pale yellow solid (m.p.116-118°C, *lit* 119-120°C); $R_f = 0.16$ (EtOAc : petrol, 1:1); IR (film): 2892, 1785, 1747, 1710, 1382, 1303, 1249, 1165, 1010, 986, 730 cm⁻¹; ¹H NMR: $\delta_{\rm H}(400 \text{ MHz}; \text{ CDCl}_3; \text{ Me}_4\text{Si})$ 1.74, 1.78 (6H, s, C(CH₃)₂), 2.13 (3H, s, COCH₃), 2.78 (2H, m, CH₂CH₂CH), 2.75 (2H, t, J 7.1 Hz, CH₂CH₂CH), 3.86 (1H, t, J 5.4 Hz, CH₂CH₂CH); ¹³C NMR: $\delta_{C}(100$ MHz; CDCl₃; Me₄Si) 19.5 (CH₂CH₂CH), 25.8, 27.9 C(CH₃)₂), 29.5 (COCH₃), 38.7 (CH₂CH₂CH), 44.0 (CH_2CH_2CH) , 104.5 $(C(CH_3)_2)$, 164.7 $(2 \times COO)$, 207.5 (CO); MS (ESI⁻): m/z 213, ([M-H]⁻, 100%); HRMS (ESI⁻) $C_{10}H_{13}O_5^{-}$ ([M-H]⁻) requires: 213.0768. Found: 213.0762.

(±) 5-(3-(Hydroxyimino)butyl)-2,2-dimethyl-1,3-diox ane-4,6-dione, 2^{27,28}

Following the general procedure 2, to the solution 6.44 mmol of Michael adduct 1 (1.38 g) in EtOH (30mL), 12.88 mmol of NH₂OH.HCl (888mg,) and 12.88 mmol of Et₃N (1.30g) was added and the mixture was gently heated (50°C) for 1 hrs, to give crude product which was purified by flash column chromatography to afford oxime 2 (1.14 g, 77%) as pale vellow semi-solid; $R_f = 0.23$ (EtOAc : petrol, 1:1); IR (film): 3319, 2982, 1786, 1743, 1715, 1372, 1307, 1240, 1176, 1012, 982, 731 cm⁻¹; ¹H NMR: (major isomer) $\delta_{\rm H}(400 \text{ MHz}; \text{ CDCl}_3; \text{ Me}_4\text{Si})$ 1.74, 1.78 (6H, s, C(CH₃)₂), 1.94, 1.91 (3H, s, CNOHCH₃), 2.31 (2H, m, CH₂CH₂CH), 2.44, 2.63 (2H, t, J 7.4 Hz, CH₂CH₂CH), 3.68 (1H, t, J 6.8 Hz, CH₂CH₂CH), 9.33 (1H, bs, OH); ¹³C NMR: $\delta_{C}(100 \text{ MHz}; \text{CDCl}_{3}; \text{Me}_{4}\text{Si})$ 13.9, 15.1 (CNOHCH₃), 21.3, 22.1 (CH₂CH₂CH), 26.2, 27.3 (C(CH₃)₂), 32.2, 35.9 (CH₂CH₂CH), 44.8 (CH₂CH₂CH), 104.8, 106.1 (C(CH₃)₂), 157.6, 158.1 (C=NOH), 162.7, 164.9 (2 × COO); MS (ESI): m/z228 ([M-H]⁻, 100%); HRMS (ESI⁻) C₁₀H₁₄NO₅⁻ ([M-H]) requires: 228.0881. Found: 228.0877.

(±) 2,8,8-Trimethyl-7,9-dioxa-1-azaspiro[4.5]dec-1ene-6,10-dione, 3^{34,35}

Following the general procedure **3**, to the solution 1.79 mmol of oxime **2** (412mg) in dry CH_2Cl_2 (20mL)

was added 5.37 mmol of Et₃N (542mg) at 0°C and followed by 2.15 mol of *p*-toluenesulphonyl chloride (410 mg) and the mixture stirred for 30 min instead tosyl oxime, isolated cyclized product as spirobicyclic pyrrolidine which was purified by flash column chromatography to afford compound 3 (297 mg, 78%) as a crystalline solid; m.p. = 98° C; R_f = 0.63 (EtOAc : petrol, 1:1); IR (film): 2933, 1782, 1731, 1452, 1170, 914, 731 cm⁻¹; ¹H NMR (400 MHz; CDCl₃; Me₄Si): $\delta_{\rm H}$ 1.78, 2.06 (C(CH₃)₂), 2.16 (3H, s, N=CCH₃), 2.59 (2H, t, J 7.6 Hz, C(4)HH), 2.97 (2H, t, J 7.6 Hz, C(3)*HH*); ¹³C NMR (100 MHz; CDCl₃; Me₄Si): δ_{C} 19.2 (COCH₃), 28.3, 28.5 C(CH₃)₂), 31.4 (C4), 41.2 (C3), 80.7 (C2), 106.2 $C(CH_3)_2$), 167.4 (2 × COO), 182.9 (C5); MS (ESI): m/z 445 ([2M+Na]⁺, 100%), 234 ($[M+Na]^+$, 95%); HRMS (ESI⁺) C₁₀H₁₃NNaO₅⁺ ([M+Na]⁺) requires: 234.0742. Found: 234.0737.

(±) 4,8,8-Trimethyl-2-phenyl-7,9-dioxa-1-azaspiro [4.5]dec-1-ene-6,10-dione, 6

Following the general procedure 3, to the solution 0.95 mmol of oxime 5 (289 mg) in dry CH_2Cl_2 (15 mL) was added 2.85 mmol of Et₃N (288mg) at 0°C and followed by 1.14 mmol of p-toluenesulphonyl chloride (216 mg) and the mixture was stirred for 30 min instead tosyl oxime, isolated cyclized product as spirobicyclic pyrrolidine which was purified by flash column chromatography to afford compound 6 (160 mg, 59%) as a crystalline solid; m.p. = 103° C; R_f = 0.46 (EtOAc : petrol, 1:1.5); IR (film): 3078, 2956, 1779, 1737, 1170, 912, 740 cm⁻¹; ¹H NMR (400 MHz; CDCl₃; Me₄Si): $\delta_{\rm H}$ 0.96 (3H, d, CH₃), 1.75, 2.00 (C(CH₃)₂), 2.55 (2H, d, J 7.5 Hz, C(4)HH), 2.79 (1H, m, C(3)H), 7.3-7.9 (ArH); ¹³C NMR (100 MHz; $CDCl_3$; Me₄Si): δ_C 28.9, 28.2 C(CH₃)₂), 31.9 (C4), 41.0 (C3), 89.7 (C5), 106.0 C(CH₃)₂), 112-139 (ArC), 167.0 (2 × COO), 180.9 (C2); MS (ESI): m/z 310 $([M+Na]^+, 95\%);$ HRMS (ESI^+) $C_{16}H_{17}NNaO_4^+$ ([M+Na]⁺) requires: 310.1055. Found: 234.1076.

(±) 8,8-Dimethyl-2-phenyl-7,9-dioxa-1-azaspiro [4.5] dec-1-ene-6,10-dione, 9

Following the general procedure **3**, to the solution 0.69 mmol of oxime **8** (200mg) in dry CH_2Cl_2 (10mL) was added 2.07 mmol of Et_3N (209mg) at 0°C and followed by 0.83 mmol of *p*-toluenesulphonyl chloride (157 mg,) and mixture was stirred for 30 min instead tosyl oxime, isolated cyclized product as spirobicyclic pyrrolidine which was purified by flash column chromatography to afford compound **9** (120 mg, 64%) as a crystalline solid; m.p. = 97°C;

R_f = 0.53 (EtOAc : petrol, 1:1); IR (film): 3093, 2931, 1778, 1735, 1177, 910 cm⁻¹; ¹H NMR (400 MHz; CDCl₃; Me₄Si): $\delta_{\rm H}$ 1.70, 2.03 (C(CH₃)₂), 2.52 (2H, t, *J* 7.4 Hz, C(4)*HH*), 2.90 (2H, t, *J* 7.4 Hz, C(3)*HH*), 7.3-7.9 (Ar*H*); ¹³C NMR (100 MHz; CDCl₃; Me₄Si): $\delta_{\rm C}$ 28.9, 28.2 C(CH₃)₂), 31.9 (C4), 41.0 (C3), 89.7 (C5), 106.0 C(CH₃)₂), 114-138 (ArC), 167.0 (2 × COO), 180.9 (C2); MS (ESΓ): *m/z* 296 ([M+Na]⁺, 95%); HRMS (ESI⁺) C₁₅H₁₅NNaO₄⁺ ([M+Na]⁺) requires: 296.0899. Found: 234.0901.

(±) 2,4,8,8-Tetramethyl-7,9-dioxa-1-azaspiro[4.5] dec -1-ene-6,10-dione, 12

Following the general procedure 3, to the solution 0.88 mmol of oxime 11 (215mg) in dry CH_2Cl_2 (10mL) was added 2.64 mmol of Et₃N (267mg) 0°C and followed by 1.06 mmol of at *p*-toluenesulphonyl chloride (201 mg,) and the mixture stirred for 30 min instead tosyl oxime, isolated cyclized product as spirobicyclic pyrrolidine which was purified by flash column chromatography to afford compound 12 (111 mg, 56%) as a crystalline solid; m.p. = 99°C; $R_f = 0.51$ (EtOAc : petrol, 1:1); IR (film) 2934, 1780, 1732, 1456, 1172, 915, 735 cm⁻¹; ¹H NMR (400 MHz; CDCl₃; Me₄Si): $\delta_{\rm H}$ 0.98 (3H, d, CH₃), 1.79, 2.05 (C(CH₃)₂), 2.16 (3H, s, N=CCH₃), 2.64 (1H, m, C(4)*H*), 2.94 (2H, d, *J* 7.5 Hz, C(3)*HH*); ¹³C NMR (100 MHz; CDCl₃; Me₄Si): δ_C 18.2, (CH₃), 21.5 (N=CCH₃), 28.4, 28.7 C(CH₃)₂), 32.1 (C4), 43.2 $(C3), 88.9 (C5), 106.7 C(CH_3)_2), 167.9 (2 \times COO),$ 182.8 (C2); MS (ESI): m/z 248 ([M+Na]⁺, 95%); HRMS (ESI⁺) $C_{11}H_{15}NNaO_4^+$ ([M+Na]⁺) requires: 248.0899. Found: 234.0839.

Conclusion

In summary, we have developed a one-pot methodology for the synthesis of spiro-heterobicyclic compounds as spiro-3,4-dihydro-2H-pyrroles with alkyl and aryl substitutions on the five member heterocycle. It was also concluded that both aryl groups as substitution was not possible as we have started with chalcone as starting material and reaction did not work. The method is useful for the synthesis of the rigid spiro-heterocycle which can also be converted to the non-natural amino acids by hydrolysis to half acid ester and interconverted to amino group through Curtius rearrangement reaction. The present methodology developed will be very useful to get quick access to spiro-3,4-dihydro-2Hpyrroles which is are part of many natural products of high importance.

Funding

Some of the work has been done at the time of Ph.D. and it was sponsored by Ministry of Social Justice and Empowerment, Government of India.

Acknowledgement

The author is very much thankful to Prof. Mark Moloney Oxford University, UK for kind support. The author also extends his gratitude to the Management of PES and Principal, and the author's family for continuous help in his research.

References

- 1 d'Ischia M, Napolitano A, Pezzella A, Trofimov B A,; Nedolya N A, Bergman J & Janosik T, in *Comprehensive Heterocyclic Chemistry III*, Vol. 3, Chapters 3.01-3.04, edited by Katritzky A R, Ramsden C A, Scriven E F V, Taylor R J K and Jones G, (Pergamon-Elsevier, Oxford), pp.1–388 (2008).
- 2 Quin L D & Tyrell J A, Fundamentals of Heterocyclic Chemistry, Chapter 9.2 (Wiley, New Jersey), pp.222–225 (2010) DOI: 10.1021/acs.joc.0c00109.
- 3 Bergman J & Janosik T, in *Modern Heterocyclic Chemistry*, Vol. 4, Chapter 4, edited by Alvaroz-Builla J, Vaquero J J and Barluenga J (Wiley-VCH, Weinheim), pp.269–375 (2011).
- 4 Trofimov B A, Mikhaleva A I, Schmidt E Y & Sobenina L N, *Chemistry of Pyrroles* (CRC Press, Boca Raton) (2015).
- 5 Lopchuk J M, in *Progress in Heterocyclic Chemistry*, Vol. 28, Chapter 5.2, edited by Gribble G W and Joule J A (Elsevier, Oxford), pp.165–218 (2016).
- 6 Vessally E A, *RSC Adv*, 6 (2016) 18619. DOI: 10.1039/C5RA20706A.
- 7 Zhou N, Zhu H T, Yanga D S & Guan Z H, Org Biomol Chem, 14 (2016) 7136. DOI: 10.1039/C6OB00962J.
- 8 Sharma A & Piplani P, *J Heterocycl Chem*, 54 (2017) 27, DOI:10.1002/jhet.2550.
- 9 Bhardwaj V, Gumber D, Abbot V, Dhiman S & Sharma P, *RSC Adv*, 5 (2015) 15233, DOI: 10.1039/c4ra15710a.
- 10 Khajuria R, Dham S & Kapoor K K, RSC Adv, 6 (2016) 37039, DOI: 10.1039/c6ra03411j.
- 11 Mohamed M S, El-Hameed R H A & Sayed A, *J Adv Pharm Res*, 1 (2017) 1; DOI:10.21608/APRH.2017.16155.
- 12 Kaur R, Rani V, Abbot V, Kapoor Y, Konar D & Kumar K, J Pharm Chem Chem Sci, 1 (2017) 17.
- 13 Kumar M R, Manikandan A, Sivakumar A & Dhayabaran V V, *Bioorg Chem*, 81 (2018) 44; DOI: 10.1016/ j.bioorg.2018.07.037.
- 14 Aldeghi M, Malhotra S, Selwood D L & Chan A W E, *Chem Biol Drug Des*, 83 (2014) 450, DOI:10.1111/cbdd.12260.
- 15 Richie T J & Macdonald S J, *Drug Discovery Today*, 14 (2009) 1011, DOI 10.1016/j.drudis.2009.07.014.
- 16 Zheng Y, Tice C M & Singh S B, *Bioorg Med Chem Lett*, 24 (2014) 3673; DOI :10.1016/j.bmcl.2014.06.081.
- 17 Santos M M M, *Tetrahedron*, 70 (2014) 9735. DOI: 10.1016/j.tet.2014.08.005.
- 18 Gubba Reddy B V, Gopal Reddy S, Durgaprasad M, Bhadra M P & Sridhar B, Org Biomol Chem, 13 (2015) 8729; DOI: 10.1039/C5OB01077B.

- 19 Lewis E J, Hunsicker L G, Clarke W R, Berl T, Pohl M A, Lewis J B, Ritz E, Atkins R C, Rohde R & Raz I, N Engl J Med, 345 (2001) 851; DOI: 10.1056/NEJMoa011303.
- 20 Massie B M, Carson P E, McMurray J J, Komajda M, McKelvie R, Zile M R, Anderson S, Donovan M, Iverson E, Staiger C & Ptaszynska A, *N Engl J Med*, 359 (2008) 2456; DOI: 10.1056/NEJMoa0805450.
- 21 Kuhl H, *Climacteric*, 8 (2005) 3; DOI:10.1080/13697 130500148875.
- 22 Wiesinger H, Berse M, Klein S, Schwend S, Hochel J, Zollmann F S & Schutt B, *Br J Clin Pharmacol*, 80 (2015) 1399; DOI: 10.1111/bcp.12745.
- 23 Claire M, Faraj H, Grassy G, Aumelas A, Rondot A & Auzou G, J Med Chem, 36 (1993) 2404; DOI: 10.1021 /jm00068a018.
- 24 Friedman A J, *Cutis*, 96 (2015) 216; DOI: 10.2147/CCID.S137794.
- 25 Bariwal J, Voskressensky L G & Van der Eycken E V, *Chem Soc Rev*, 47 (2018) 3831; DOI: 10.1039/C8OB03122C.

- 26 Iwata A, Inuki S, Oishi S, Fujii N & Ohno H, *Tetrahedron*, 71 (2015) 6580; DOI: 10.3987/COM-19-S(F)40.
- 27 Chandan N, Thompson A L & Moloney M G, Org. Biomol Chem, 10 (2012) 7863; DOI: 10.1039/c2ob26423d.
- 28 Wahyuningsih T D, Pchlk K, Kumar N & Black D, *Tetrahedron*, 62 (2006) 6343.
- 29 Sannigrahi M, *Tetrahedron*, 55 (1999) 9007; DOI: 10.1016/S0040-4020(99)00482-2.
- 30 Chen B C, *Heterocycles*, 32 (1991) 529; DOI: 10.3987/REV-90-422.
- 31 Sapi J & Laronze J Y, ARKIVOC, (vii) (2004) 208-22; DOI : 10.3998/ark.5550190.0005.717.
- 32 Tomori H, Shibutani K & Ogura K, Synthesis, 224 (1984) DOI: 10.1055/s-1984-30779.
- 33 Renslo A R & Danheiser R L, J Org Chem, 63 (1998) 7840; DOI : 10.1021/jo026258k.
- 34 Bonvallet P A, Todd E M, Kim Y S & McMahon R J, J Org Chem, 67 (2002) 9031; DOI: 10.1021/jo020304z.
- 35 Singh V, Kanojiya S & Batra S, *Tetrahedron*, 62 (2006) 10100; DOI: 10.1016/j.tet.2006.08.045.