



Synthesis and (some) applications of carbon-nanotube-supported pyrolytic carbon nanocones

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Graphene-based cones with nanosized apex can be obtained by means of a high temperature pyrolytic carbon deposition process using methane and hydrogen as gaseous feedstock and single carbon nanotubes as deposition substrates. Aside the cones, micrometer-sized carbon beads or fibre segments are deposited meanwhile which are a key morphological component for allowing handling and mounting the carbon cones and then using them for various applications. Based on both the literature dealing with pyrolytic carbon deposition processes and experimental observations, a peculiar deposition mechanism is proposed, involving the transient formation of pitch-like liquid phase droplets which deposit onto the individual carbon nanotubes. In this picture, it is believed that a key parameter is the diameter ratio for the droplets and the nanotubes, respectively. The cone concentric texture and perfect nanotexture are shown by high resolution transmission electron microscopy, which allows interesting mechanical and conducting properties to be predicted. Correspondingly, applications of the carbon nanocones as electron emitters for cold-field electron sources on the one hand, and as probes for various modes of near-field microscopy on the other hand, have been tested.

Keywords: Micro-/nano-cones, Growth mechanisms, Pyrolytic carbon, Near-field microscopies, Electron emitters

1 Introduction

All-graphene carbon cones with nano-sized apex can be obtained by means of a high temperature pyrolytic carbon deposition process using methane and hydrogen as gaseous feedstock and individual carbon nanotubes (CNTs) as deposition substrates^{1,2}. Aside the cones with smooth surface, micrometer-sized carbon beads (Figs. 1b and 1c) or fiber segments (Figs. 1a and 1c) are deposited meanwhile, which are a key morphological component for allowing handling and mounting the carbon cones and then using them for various applications.

2 Synthesis and growth mechanisms

Carbon nanotubes (CNTs) are grown first, using iron nanoparticles as catalyst combined with a regular chemical vapor deposition process (CCVD) in which a $\text{CH}_4 + \text{H}_2$ feedstock is cracked at 1000-1100 °C². Then catalyst-free conditions (CVD) involving temperatures in the 1300-1500 °C range and a modified H_2/CH_4 ratio are used to deposit pyrolytic

carbon onto each CNT, which spontaneously adopt the complex morphologies as shown in Fig. 1, while aligning more or less periodically along the CNTs¹⁻⁴ and showing a high nanotextural quality (see Fig. 2). A deposition mechanism which requires both the presence of small carbon radicals and the transient formation of pitch-like (i.e., containing polyaromatic hydrocarbon molecules (PAHs) and lighter hydrocarbons) droplets in the gas phase was proposed, based on both the literature and experimental evidences⁴ (Fig. 3).

At the beginning of the deposition process, radicals cannot chemisorb onto the CNTs because of the lack of active sites. Therefore, based on observations of Monthioux *et al.*⁴, the proposed mechanism for the early deposition step is related to the wettability of the substrate (the primary nanotube surface) by the carbon species (PAH-containing droplets) where the deposition and carbonization processes take place. Because of the substrate geometry (non-planar), two situations of partial wetting define the way droplets are deposited onto the nanofilament. These situations depend on the nanotube over droplet diameter ratio.

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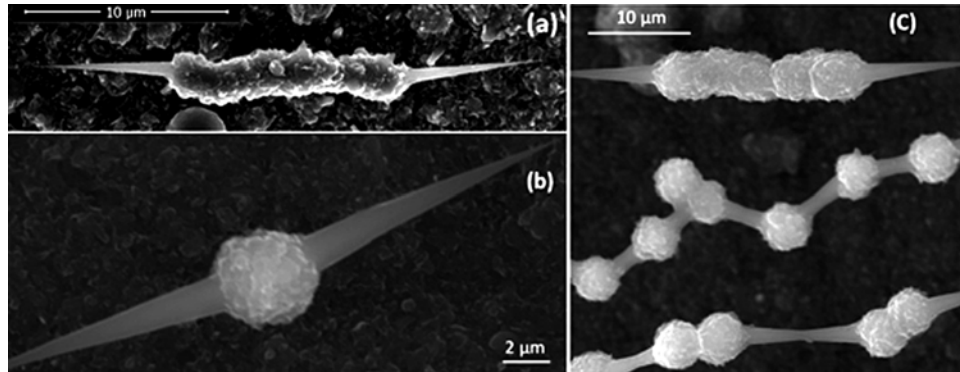


Fig. 1 — (a) Morphology of a "carbon nanocone" with a fiber segment, (b) Morphology of a "carbon nanocone" with an isolated carbon bead and (c) Co-existence of different "carbon nanocone" morphology types: chains of carbon beads linked by truncated cones, and a fiber-segment/cone morphology.

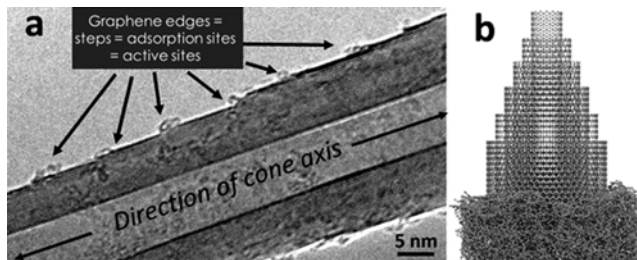


Fig. 2 — Carbon nanocone texture and nanotexture (a) High resolution transmission electron microscopy (HRTEM) image showing a perfect orientation of the nanocone graphene layers with respect to the cone axis (primary carbon nanotube). As indicated (see arrows) the cone surface is made of graphene edges, each of them being a preferred site for adsorption and (b) Display of the constituting graphenes within the conical part (model by I. Suarez-Martinez, Curtin University, Australia).

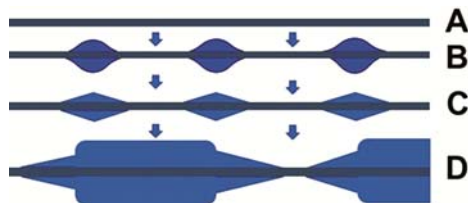


Fig. 3 — Sketch of the formation mechanisms of the carbon nanocone-bearing morphologies seen in Figure 1. A: growth of the primary CNTs; B: deposition of pitch-like (PAH-containing) droplets onto the CNTs, forming double meniscus; C: PAHs align parallel to the CNT axis upon surface tension forces and then subsequently carbonise, forming bicones; D: Growth of the cones mostly from C radicals, while the short fiber segments (or beads) form mostly from droplets.

Should the droplets exhibit diameters smaller than the CNT diameters (5-10 nm), they would deposit on the filament surface as they would onto a planar surface, corresponding to the first situation. The second situation occurs when the carbon nanotubes have smaller diameters than droplets. In such a case, each PAH-rich droplet is deposited onto the carbon

nanotube surface forming a double meniscus (Fig. 3b). The primary droplet deposition event is assumed to be somewhat periodic following the Plateau-Rayleigh instability⁵. These double menisci are responsible for forming the primary double nanocones observed in short duration experiments (step C in Fig. 3)⁴. Indeed, The PAHs from the liquid double meniscus align parallel to the nanotube surface because of the capillary forces and surface tension and then carbonise as concentric graphenes (step C in Fig. 3). The cones continue growing the same way until a threshold value of about 100 nm for the diameter at the cone base is reached. Once the threshold value is reached, the deposition of the droplets continues at the base of the double cones now following situation 1 above-mentioned, because at that time, the radius of curvature of the substrate surface has become long enough for the droplets to deposit on it as they would do onto a flat substrate. From this moment, the graphenes resulting from the carbonization of the PAHs no longer align well, because the effect of surface tensions no longer prevails, producing morphologies with rough surface. Rough surface sub-morphologies can be bead-like or fibre-segment-like (Fig. 1) probably depending on the number of droplets in the gas phase. Once this step is reached, the cones and the beads/fibre segments continue to grow simultaneously, but now driven by a competition between the two carbon deposition modes: (i) the chemisorption of radicals onto the active sites present at the cone surface and (ii) the physisorption of PAH-rich droplets mostly onto the bead/fibre-segments, respectively.

3 Application of carbon cones as electron emitters

The concentric texture and the graphene perfection of the carbon cones are likely to provide them with the

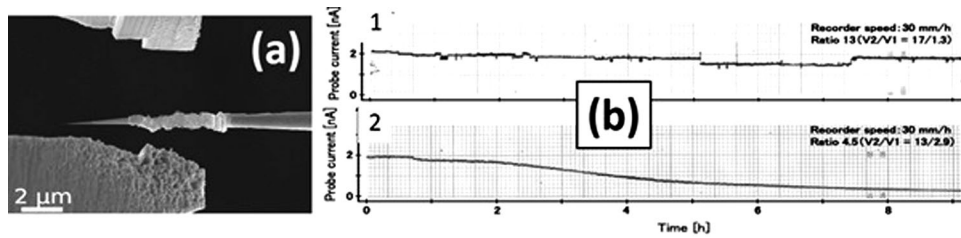


Fig. 4 — (a) Carbon cone W-welded by focused ion beam onto a regular yet truncated W tip emitter and (b) comparison of the current stability with time for (b-1) the carbon cone emitter; (b-2) a regular W tip emitter. Such a stability is unprecedented for CFEG electron source⁸.

physical properties of large multiwall CNTs, typically, high electrical and thermal conductivities, high current density, and high flexural resilience, yet with an apex in the ~5-10 nm range only. Therefore, carbon cones were successfully tested as electron emitters instead of the usual W emitter, in cold-field emission guns (CFEG) (Fig. 4a), the best current electron sources for electron microscopes. Results overpassed the expectations, and the carbon cone emitter revealed to be able to provide 5 times higher reduced brightness and better coherence than any commercial electron source for TEM⁶⁻⁸. They also are better than pristine CNTs, because of the conical shape which provides sufficient mechanical stability to prevent the tip from vibrating while emitting. Overall, the carbon cone electron source exhibits much better time stability than the best CFEG on market (Fig. 4b). Carbon cones were then patented for this application⁹.

4 Application of carbon cones as probes for scanning probe microscopy

The aspect ratio and electrical conductivity property (which is expected from the perfect nanotextural quality, see Fig. 2) make the carbon cones presumably suitable as probes for various modes of near field microscopy. The carbon nanocone shape have a length L in the range of (2-7 μm) and a base-width W of about up 700 nm, and a very sharp apex angle noted 2α . The smaller the cone radius $W/2$, the closer to zero the cone angle. This very small angle can be estimated by the following equation:

$$\text{Tan.}\alpha = \alpha = W/(2 * L) \quad \dots (1)$$

this leads to:

$$2\alpha = W/L \quad \dots (2)$$

For $W = 700 \text{ nm}$; $L = 6 \text{ }\mu\text{m}$ (maximum value) as depicted in Fig. 5a, the cone angle 2α takes the value of 0.12° .

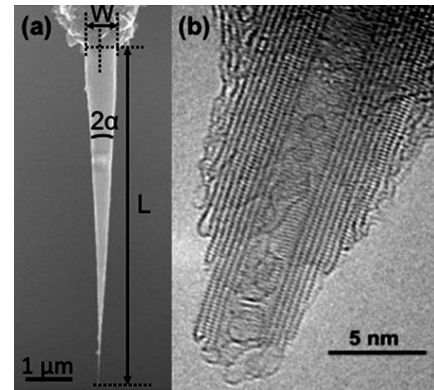


Fig. 5 — Main geometrical characteristics of a nanocone tip (a) SEM image showing the length, width and angle parameters and (b) High resolution transmission electron microscopy (HRTEM) image of the cone apex. (Photo credit: R. Wang, CEMES-CNRS).

Carbon cone morphologies such as shown in Fig. 1a were glued onto tailor-made AFM doped-Si cantilevers (obtained from AppNano) following either a micro-manipulation procedure under an optical microscope as initially proposed by Dai *et al.* for CNTs¹⁰, or the same focused electron beam-induced welding techniques as carried-out for mounting the cones as electron emitters^{6,7}. In the latter case, the mounting process was conducted using a Scanning Electron Microscope (SEM Zeiss 1540 XB) equipped with a dual Focused Ion Beam (FIB) / Electron Beam system and a micromanipulator (clamps) as depicted in Fig. 6. Subsequently, tests of the probes for different conductive and non-conductive SPM modes were conducted.

Carbon nanocones exhibit better resolution than standard AFM Si tips when used for AFM-tapping mode topography images (Fig. 7). The image parameters of a scanned area of 500 nm in width with a scan speed of 2hz and 256 X 256 scan pixels allowed distinguishing the grainy surface of the silicon wafer background as well as the separation between two gold nanoparticles which are clearly delimited thanks to the

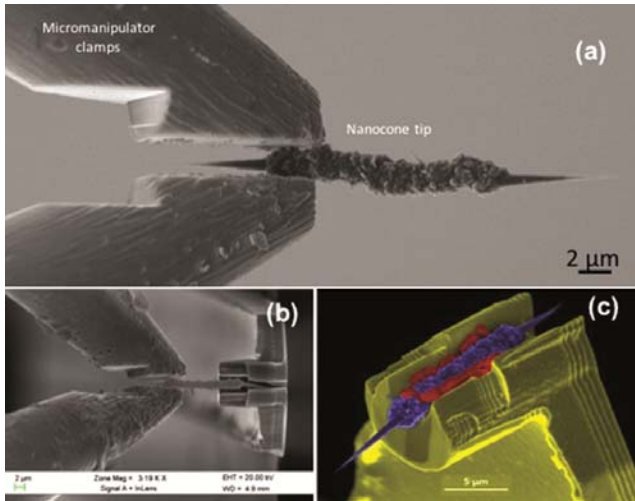


Fig. 6 — Mounting a carbon nanocone tip as SPM probe by focused electron-beam-induced metal deposition. (a) A carbon nanocone morphology is grabbed by micromanipulator clamps and (b) The carbon nanocone morphology is placed into the groove of a specially-designed doped-Si cantilever; (c) The carbon nanocone tip (welded onto the cantilever) is ready to be used for different AFM tests: the yellow colour represents a doped-Si cantilever, the purple colour, the carbon cone tip morphology, and the red parts correspond to the platinum metal deposit used for welding.

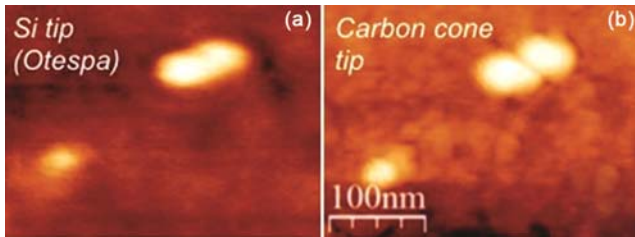


Fig. 7 — (a) AFM image obtained from a commercial Si probe on gold nanoparticles deposited on a Si wafer and (b) Same area as (a) using a carbon cone probe.

tip sharpness of the carbon cone tip (Fig. 7b). Thanks to their graphenic nature, these conical tips are also suitable for near-field modes requiring conducting probes, such as conducting AFM, charge injection, and Kelvin force microscopy, for which they have already shown a promising behaviour¹¹. Other conductive modes such as C-AFM, Scanning Tunnel Microscopy (STM), and non-conductive ones such as Peak-Force Quantitative Nanomechanical AFM (PFQNAFM) are currently tested, and work in this area is in progress.

5 Conclusions

In summary, we have shown that unique carbon conical morphologies can be produced from pyrolytic carbon deposition onto nanoscale substrates (carbon nanotubes) during a thermal CVD process. The fact

that micro-nano morphologies are formed while made up of both smooth nanocones as well as rough bead or fibre segments indicates (at least under experimental conditions of high temperature and atmospheric pressure) a pyrolytic carbon deposition process driven by two different mechanisms. This can be explained by the coexistence (in the gas phase) of carbon radicals, which chemisorb on the cone surface because it exhibits the needed active sites, and of PAH-containing droplets which randomly deposit anywhere.

On the other hand, it was demonstrated that the well-controlled geometrical characteristics (i.e. aspect ratio, tip radius) of the synthesized morphologies, and their excellent physical properties (i.e. good electrical conductivity, mechanical strength) as well as their relatively easy mounting, allow them to have the potential to equip electron sources with unprecedented performances, and to be unique, presumably durable, multi-mode probes for various kinds of near-field microscopy.

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