

Properties of PS/TiO₂ electrospun fibres using limonene as a solvent

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Limonene, a natural solvent, has been used for producing polystyrene (PS) nanocomposite (NC) fibres. Nanocomposite fibres of PS are prepared by electrospinning of a homogeneous solution of titanium dioxide (TiO₂) nanoparticles (NPs) and PS. X-ray diffraction (XRD) pattern of PS nanocomposite fibres confirms the presence of TiO₂ nanoparticles in the samples. FTIR spectra of PS nanocomposite fibres obviously show that there is no chemical linkage or interaction between PS and TiO₂ nanoparticles in the resulting composite fibres. The morphology of PS electrospun fibres and PS/TiO₂ nanocomposite fibres is investigated by SEM and FE-SEM. FE-SEM images of electrospun fibres reveal some aggregation of TiO₂ nanoparticles. The results also show that increasing TiO₂ nanoparticles reduces PS electrospun fibres diameter. Also, the UV protection of PS electrospun fibres is enhanced due to the increase in TiO₂ nanoparticles load. Tensile strength and elasticity modulus first show an increase up to 4 wt% of TiO₂ and then a decrease at higher loading. Differential scanning calorimeter (DSC) thermograms for PS electrospun fibres indicate that the introduction of TiO₂ nanoparticles decreases the glass transition temperature (T_g).

Keywords: Electrospinning, Limonene, Nanofibres, Nanoparticles, Polystyrene

1 Introduction

Nanotechnology has been well developed during the past two decades with a significantly growing number of studies on nanofibres and their applications¹⁻⁴. In particular, the nanofibres and nanomaterials science in textiles have received much attention, showing various applications. Electrospinning as a simple and inexpensive technique is used for producing fibres with a diameter ranging from submicrometers to nanometers. Electrospun fibre mats, due to their high surface area, small pore size distribution and high porosity, are good candidates that can be employed in many applications⁴.

Recently, the use of various types of nanoparticles dispersed in polymer solutions have attracted great interest in nanofibres production because they often show dramatic improvement in the properties of electrospun fibres as compared to virgin fibres⁵.

Amorphous polystyrene (PS) is a transparent and commonly used plastic. It is a hard and brittle polymer with very high electrical resistance and low dielectric loss⁶. This versatile polymer, which has several advantages such as no pollution and toxicity, is used in various fields⁶. Electrospinning of PS has

also attracted great interest because of its varied applications in areas such as ion exchange^{7,8}, filtration⁹, tissue engineering¹⁰, etc. Numerous researchers have investigated the structure and properties of electrospun PS fibres¹¹⁻¹⁸. Common organic solvents such as N, N-dimethylformamide (DMF), tetrahydrofuran (THF), N, N-dimethylacetamide (DMAc), toluene, etc. are well known as PS solvents in electrospinning¹⁹⁻²³. These solvents are not environment-friendly because of their toxicity and may emit an unpleasant smell or toxic gases. García *et al.*²⁴ tested several natural solvents as dissolution agents for extruded polystyrene. Noguchi *et al.*²⁵, by using d-limonene as a green solvent, developed a method to shrink expanded polystyrene (EPS)²⁵. Shin *et al.*²⁶ produced PS nanofibres with EPS solution using d-limonene²⁶. They compared electrospun PS fibre from EPS solution dissolved in d-limonene with PS nanofibre produced from dissolving PS in other solvents such as THF, DMAc, and DMF. It was observed that although d-limonene was a good solvent for polystyrene, at low concentrations, the electrospinning jet stability of PS was poor compared to other solvents.

D-Limonene, a monoterpene hydrocarbon, is a natural and biodegradable solvent. It is the principal component of orange or lemon peel oil^{26,27}. As a

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solvent, DL-limonene can be used to replace toxic, hazardous, and dangerous organic solvents. D-Limonene has various valuable usages including medical and pharmaceutical products, perfumes, flavorings, pesticides, cleaners, heat transfer fluids, and solvents^{26,27}.

Electrospinning process of PS is mostly achieved using organic solvents. It has some problems such as environmental problem, difficulty of process handling, solvent costs, and presence of the trace of solvent impurities.

Recently, the use of various types of nanoparticles (NPs) dispersed in PS has been extensively studied to enhance the properties of electrospun PS and consequently, extend its application fields²⁸⁻³².

Mazinani *et al*²⁸ investigated the morphology and properties of PS/carbon nanotube (CNT) electrospun fibres. They used styrene-butadiene-styrene type as an interfacial agent to modify the dispersion of CNTs. The results showed that fibres diameter was decreased by increasing CNT concentration. Also, improvement in electrical conductivity and mechanical properties was obtained by the addition of CNT below percolation. Preparation of bead-free PS/Ni electrospun fibre with higher thermal stability was reported by Chen *et al*²⁹. Rojas *et al*³⁰ revealed that by using cellulose nanowhiskers as the reinforcing material in the electrospinning of PS, the elastic modulus of electrospun fibres was increased with nanowhisiker content. Also, T_g of electrospun fibres tended to decrease with increasing cellulose nanowhiskers load. Kim *et al*³¹ produced PS/gold nanocomposite (NC) fibres³¹. Based on the results, an increase in gold nanoparticles load resulted in an increase in the surface roughness of electrospun PS/gold fibres and subsequently, a decrease in the diameter and T_g of electrospun fibres. Electrospinning of PS containing zinc oxide and titanium dioxide nanoparticles was reported by Kobayashi *et al*³². They found that the morphology of electrospun fibres was changed by the addition of nanoparticles in PS solutions.

In this study, DL-limonene as a green media was used as a solvent for PS. PS/TiO₂ nanocomposite fibres with different TiO₂ nanoparticles loadings were electrospun. The effect of the concentration of TiO₂ nanoparticles on the properties of PS electrospun fibres was investigated. The morphology of the electrospun nanocomposite fibres was also explored using scanning electron microscopy. In addition, UV

absorbance spectra, and tensile and thermal properties of the resulting electrospun mats at different TiO₂ loads were characterized by using UV-vis spectrophotometer, Zwick 1446-60 tensile tester and differential scanning calorimeter respectively.

2 Materials and Methods

Polystyrene (PS) pellets with the weight-average molecular weight (M_w) of 280,000 g/mol and TiO₂ nanoparticles with a particle size of about 25 nm and a specific surface area of about 200-220 m²/g were received from the Sigma Aldrich. DL-Limonene was purchased from Merck as the solvent. Ethanol was purchased from Merck as the co-solvent. LiCl was one of the commercial products of Pinjiang Chemical Co.

2.1 Preparation of Polymer Solution

A polymer solution with the concentration of 35wt. % was prepared by dissolving PS in a mixture of DL-limonene/methanol (90:10). LiCl (1.5%), as a salt additive, was added into the polymer solution to improve electrical conductivity. The slurry was prepared by dispersing the predetermined amounts of TiO₂ nanoparticles into DL-limonene solvent using ultrasonic vibration for 10 min until the nanoparticles were uniformly dispersed in the solvent. This dispersion of TiO₂ was added into the PS solution and vigorously stirred at ambient temperature for 12h. The compositions of TiO₂ to PS were 0, 2, 4, 6, and 8 wt % and these PS/TiO₂ nanocomposites were referred to as PS-0, PS-2, PS-4, PS-6, and PS-8 respectively.

2.2 Electrospinning Process

The polymer solution was loaded into a glass syringe with a 23 gauge needle. The syringe was fixed horizontally on a syringe pump (Model: STC-527, Terumo Co), in which the polymer solution feed rate and traverse speed of pump were fixed at 0.3 mL/h and 4 cm/min respectively. A 20 kV voltage was applied between the needle and a metallic collector. The tip-to-collector distance of 18 cm was used. Electrospun fibre mats were collected on the rotating drums with the diameter of 15 cm and the rotation speed of 70 rpm. Electrospinning was performed at room temperature for 3h.

2.3 Characterization

Scanning electron microscopy (SEM) of Seron Technology AIS-2100 and field emission scanning electron microscope (FESEM) of Hitachi S-4160 were used to characterize the morphology of

electrospun fibres. At least 100 fibres were used to calculate the mean values of the diameter of electrospun fibres. The fibre diameter distribution was evaluated by a program created in MATLAB 7.

A wide angle X-ray diffraction (WAXD) analysis was performed on samples using a Philips Xpert-MPD X-Ray diffractometer with CuK α radiation

($\lambda=1.54 \text{ \AA}$). Scans were run in the wide angle of 5–80° and scan speed of 1°/min to confirm the presence of TiO₂ nanoparticles in the fibres and study the crystal form of nanoparticles and fibres in the presence of TiO₂. The fourier transform infrared (FTIR) spectra were obtained on a Gasco 680 plus FTIR spectrometer in the range of 400 - 4000 cm⁻¹ by KBr disk method.

The UV absorbance spectra of PS and PS/TiO₂ electrospun mats were recorded by UV–vis spectro photometer (Gasco V-750) in the wavelength range 200 - 800 nm. The sample dimensions were 1× 4 cm (width and length). The thickness of the samples was varied from 120 μm to 140 μm .

The mechanical properties were measured by Zwick 1446–60 tensile tester at the standard conditions. Sample dimensions were 5 mm in width and 15 mm in length with the cross head speed of 5 mm/min. The samples were placed under standard atmospheric conditions (20 \pm 2°C and 65 \pm 2 % RH) for 24 h before the experiments. Differential scanning calorimeter (DSC) was carried out using TA instrument under a nitrogen atmosphere and a heating rate of 10 °C/min to measure glass transition temperatures (T_g).

3 Results and Discussion

3.1 Morphology of Fibres

The SEM images and diameter distribution diagrams of electrospun PS fibres and PS/TiO₂ nanocomposite fibres are shown in Fig.1. The average diameters of PS and PS/TiO₂ electrospun fibres are listed in Table 1. It is observed that the morphology and diameter of PS electrospun fibres are significantly affected by TiO₂ content. The average of PS/TiO₂ fibres diameter is varied from 1.73 μm to 2.09 μm . SEM micrographs of PS and PS/TiO₂ fibres show that PS microfibres are uniform without any beads. Results also show that the average diameter of PS/TiO₂ nanocomposite fibres is decreased by increasing TiO₂ nanoparticles. The TiO₂ nanoparticles are semiconductor³⁵ and the addition of TiO₂ nanoparticles increases the electrical conductivity of

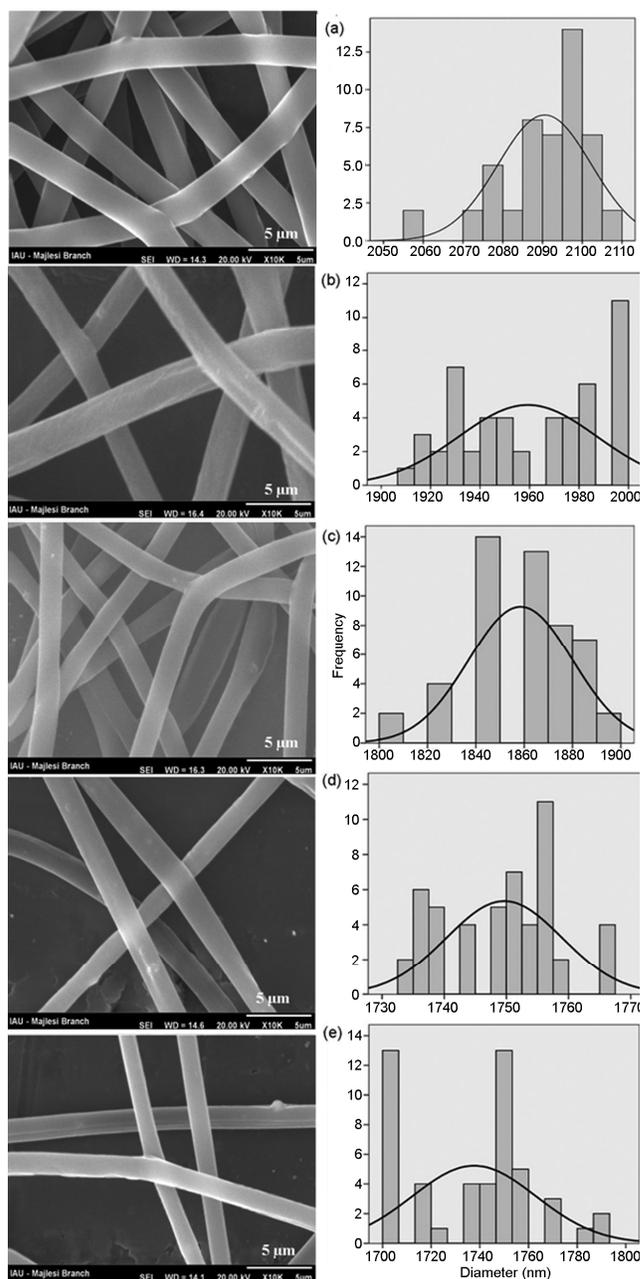


Fig.1—SEM images and nanofibres diameter distribution curves of PS/TiO₂ NC fibres at different TiO₂ contents (a) pure PS, (b) 2% TiO₂; (c) 4% TiO₂; (d) 6% TiO₂, and (e) 8% TiO₂

Table 1—Effect of TiO₂ content on PS electrospun NC fibres diameter and T_g

Sample	TiO ₂ content wt %	Diameter μm	T_g °C
PS-0	0	2.09 (1.50)	87
PS-2	2	1.95 (1.42)	-
PS-4	4	1.85 (2.76)	80
PS-6	6	1.74 (3.43)	-
PS-8	8	1.73 (2.56)	79

Values in parentheses show the coefficient of variation.

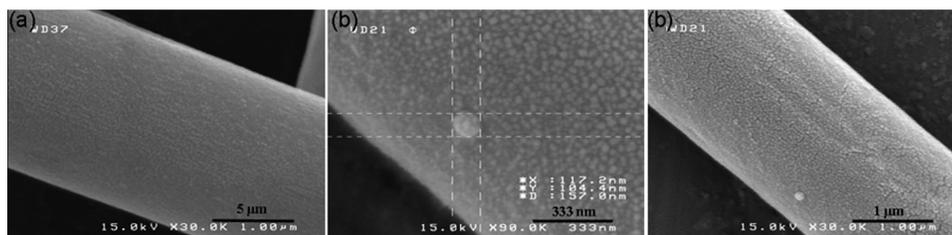


Fig.2—FE-SEM images of (a) pure PS; and (b) PS/TiO₂ NC fibres including 8% wt TiO₂NPs

Table 2—Duncan's multiple-rang test for data of Table 1

Sample	N	Subset for alpha = 0.05			
PS-8	50	1.73	-	-	-
PS-6	50	1.74	-	-	-
PS-4	50	-	1.85	-	-
PS-2	50	-	-	1.95	-
PS-0	50	-	-	-	2.09
Sig.	-	0.486	1.000	1.000	1.000

polymer solution and cause the reduction of PS electrospun fibres diameter. The statistical analysis (Table 2) shows that when TiO₂ nanoparticles content is increased from 0 to 6%, decrease in nanofibres diameter is significant, but when 8% TiO₂ nanoparticles is loaded, decrease in nanofibres diameter is not significant in comparison to nanofibres containing 6% TiO₂ nanoparticles.

The FE-SEM has been used to evaluate the dispersion of TiO₂ nanoparticles on the surface of PS nanocomposite fibres. The FE-SEM images of pure PS fibres and PS/TiO₂ nanocomposite fibres containing 8%wt TiO₂ nanoparticles can be observed in Fig. 2. These images show some agglomeration of TiO₂ nanoparticles on the surface of electrospun fibres (Fig. 2b). It can be observed from Fig. 2(b) that the size of some TiO₂ nanoparticles is about 117 nm. The aggregation of TiO₂ nanoparticles is due to the hydrophobic nature of PS and there is not enough steric hindrance³⁴.

3.2 X-ray Diffraction Analysis

The wide angle X-ray diffraction patterns of TiO₂ nanoparticles, PS electrospun fibres and PS/TiO₂ nanocomposite fibres containing 8 wt% nanoparticles are shown in Fig. 3. TiO₂ nanoparticles show two diffraction peaks at $2\theta=25^\circ$ and 48° , which belong to (101) and (200) plane reflection of TiO₂ respectively, thereby indicating that TiO₂ is in the anatase phase³⁵. A broad diffraction peak at about $2\theta=20^\circ$ is observed in PS fibres corresponding to the PS noncrystalline phase. As indicated in Fig. 3c, apart from the diffraction peak of PS, only one of the diffraction peaks at about $2\theta=26^\circ$ can be indexed as the anatase

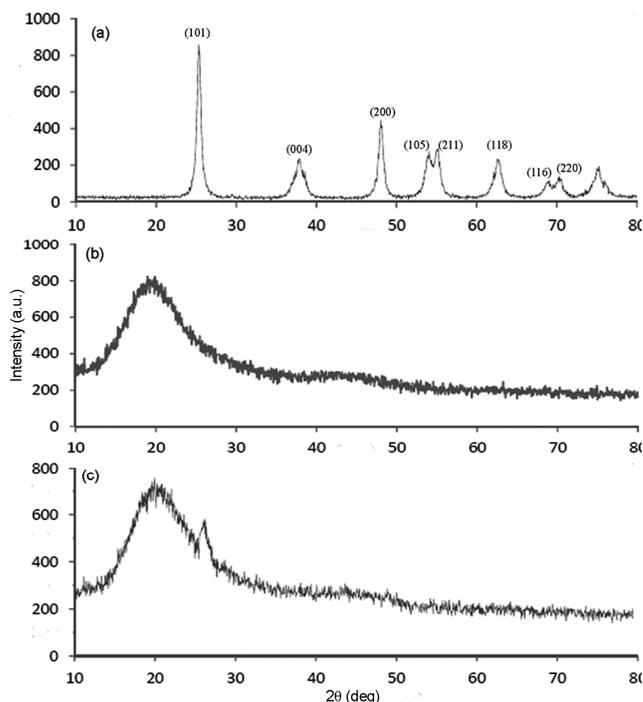


Fig.3—X-ray diffraction spectra of (a) TiO₂ NPs, (b) electrospun PS fibres, and (c) electrospun PS fibres containing 8 wt% TiO₂NPs

phase of TiO₂ for PS/TiO₂ nanocomposite fibres loaded with 8 wt. % TiO₂. This observation can be attributed to the small amount of TiO₂ in the PS fibres.

3.3 FTIR Spectroscopy

In order to evaluate the chemical interaction between the PS and the TiO₂ nanoparticles, FTIR measurements are performed. Figure 4 shows FTIR spectra of TiO₂, PS electrospun fibres and PS/TiO₂ nanocomposite fibres. The characteristic peak of TiO₂ nanoparticles is a strong kind of absorption peak that appears at about 533 cm⁻¹. FTIR spectrum of PS (Fig. 4b) shows several characteristic peaks, such as the aromatic C=C stretching vibration peaks at 1400-1600 cm⁻¹, 700 and 750 cm⁻¹ (monosubstituted benzene). The absorption bands at 3083, 3061 and 3026 cm⁻¹ are all associated with unsaturated C-H stretching vibration in the phenyl ring of PS. The bands assigned to C-H bending vibration appear at

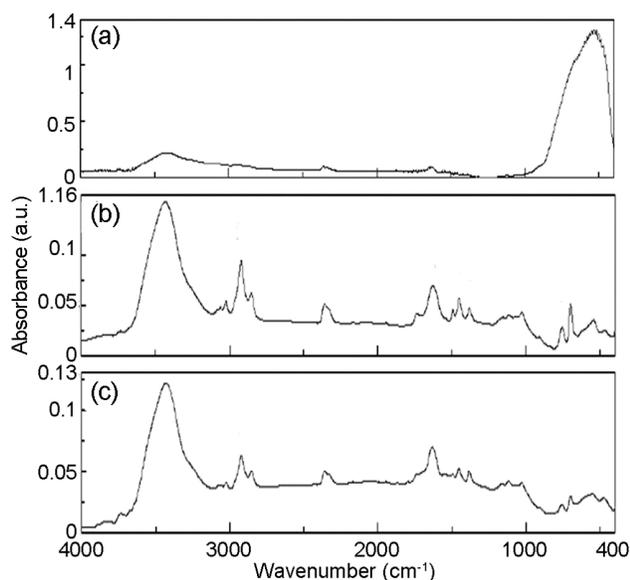


Fig.4—FTIR spectra of (a) TiO₂ NPs, (b) PS, and (c) PS/TiO₂ NC fibres with 8 wt% TiO₂

1452 and 1493 cm⁻¹. The peaks at 755 and 698 cm⁻¹ show the C–H stretches in the monosubstituent of the aromatic ring. No significant differences between the FTIR spectra for both materials are noticed. Pure PS and PS/ TiO₂ nanocomposite fibres present the same FTIR spectra. The addition of TiO₂ nanoparticles does not shift or change absorption band in electrospun nanocomposite fibres. This indicates that there is no chemical interaction between PS and TiO₂ nanoparticles in the resulting composite fibres. Also, since the amount of nanoparticles in the sample is negligible with respect to polymer, no peak from nanoparticles in nanocomposite spectrum is observed.

3.4 Thermal Analysis of Fibres

Figure 5 shows the DSC thermographs of electrospun PS and PS/TiO₂ nanocomposite fibres. Table 1 compares the results of glass transition temperatures (T_g) of these samples as determined from the DSC thermographs. These results show that T_g of PS fibres is decreased as the concentration of TiO₂ in PS is increased. The T_g value could be attributed to the polymer chains mobility. So, the decrease in T_g value of PS/TiO₂ fibres by the increase in TiO₂ nanoparticles content is due to the enhancement of the mobility of polymer chains. In other words, TiO₂ nanoparticles act as plasticizers. In fact, plasticizers work by embedding themselves between the chains of polymers, spacing them apart (increasing the free volume), and as a result, lowering the glass transition temperature for the plastic, thus making it softer. So,

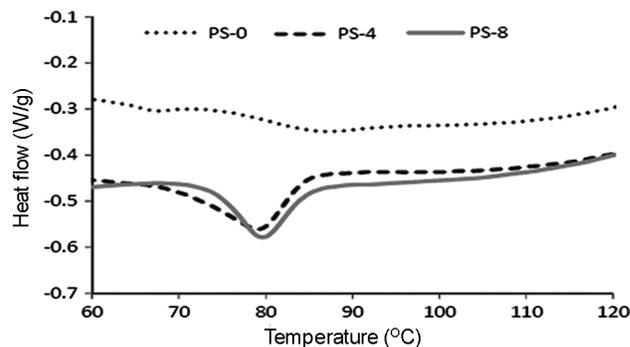


Fig.5—DSC curves of PS/TiO₂ electrospun fibres with different NPs contents

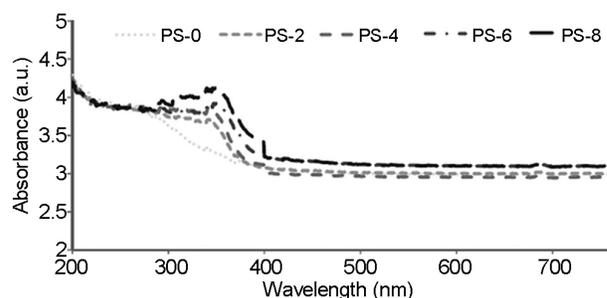


Fig.6—Absorbance spectra in the UV-vis of pure PS and PS/TiO₂ electrospun fibres

it can be said that these nanoparticles decrease T_g of PS/TiO₂ electrospun fibres by increasing the free volume of polymer. The same results are reported by Kim *et al.*³¹ for PS/Au nanocomposite fibres.

3.5 UV-vis Spectra of Electrospun PS/TiO₂ Nanocomposite Fibres

Figure 6 shows the absorbance spectra of PS and PS/TiO₂ electrospun fibres in the ultraviolet and visible region (190- 700 nm) of electromagnetic spectrum. The addition of TiO₂ nanoparticles increases the absorbance values at all wavelengths of UV region. Based on the results, the maximum absorption by samples is located at about 340 nm, which is shifted to higher wavelengths by the increase in TiO₂ content. For this point, the absorption is decreased by wavelength increase at 400 nm. As a result of loaded TiO₂ nanoparticles, the UV protection of PS electrospun fibres is increased. UV protection of PS/TiO₂ nanocomposite fibres is due to the scattering of UV rays through the high refractive index of TiO₂ and/or absorption of UV rays because of the semiconductive properties of TiO₂³⁶.

3.6 Mechanical Properties of Electrospun PS/TiO₂ Nanocomposite Fibres

Several parameters such as particle-matrix interface adhesion, particle size, particle dispersion

Table 3—Mechanical properties of PS and PS/TiO₂ electrospun mats

Sample	Tensile strength MPa	Modulus MPa	Elongation at break %
PS-0	1.37 ± 0.73	41.77 ± 28.7	1.69 ± 0.57
PS-2	1.9 ± 0.51	67.21 ± 21.48	2.85 ± 0.65
PS-4	3.26 ± 0.49	93.36 ± 15.41	2.76 ± 1.08
PS-6	2.48 ± 0.37	64.64 ± 18.17	3.87 ± 0.7
PS-8	2.05 ± 0.29	60.73 ± 15.17	3.28 ± 0.78

Table 4—Duncan's multiple-rang test for tensile strength and modulus data of Table 3

Sample	N	Subset for alpha = 0.05				
Tensile strength						
PS-0	5	1.3700				
PS-2	5		1.9000			
PS-8	5			2.0500		
PS-6	5				2.4800	
PS-4	5					3.2600
Sig.		1.000	1.000	1.000	1.000	1.000
Modulus						
PS-0	5	41.7700				
PS-8	5		60.7300			
PS-6	5			64.6400		
PS-2	5				67.2100	
PS-4	5					93.3600
Sig.		1.000	1.000	1.000	1.000	1.000

and particle loading influence the mechanical properties of composites³⁷. The variations in tensile properties of PS and PS/TiO₂ electrospun mats with TiO₂ content are listed in Table 3. As can be seen, the tensile strength and modulus of electrospun mats are increased with the increase in nanoparticles content up to 4 wt% above, showing a tendency to decrease with the increase of nanoparticles loading. With the high amount of nanoparticles above 4 wt%, the tensile strength and modulus decrease. This could be due to the aggregation of nanoparticles. As shown in Table 4, the increase in TiO₂ nanoparticles content makes significant difference in the strength and modulus of nanofibre mats. Tensile strength of nanocomposites can be enhanced when interfacial adhesion is improved. This can be correlated to the interaction of filler with the matrix. Therefore, well-adhering nano-TiO₂ can bear on the part of the load applied to the matrix and contributes to the tensile strength of the nanocomposites³⁸.

4 Conclusion

SEM images show that the PS fibre diameter decreases by increasing TiO₂ nanoparticles content.

FE-SEM images reveal the aggregation of TiO₂ nanoparticles in the PS nanocomposite fibres with 8%wt nanoparticles. Results obtained from X-ray diffraction of electrospun fibres bearing nanoparticles confirm the presence of nanoparticles in the samples. Investigating the FTIR spectrum of nanoparticles, pure PS and PS/TiO₂ nanocomposite fibres also show no change in the position of PS peaks or creation of a new peak in nanocomposite spectrum. This analysis indicates that there is only physical mixing of polymer and nanoparticles. Evaluating the effect of the amount TiO₂ nanoparticles on the mechanical properties of electrospun webs show that the increase in tensile strength and modulus with the increase of nanoparticles value is followed by a decrease beyond 4 wt%. Furthermore, considering the differential scanning calorimeter of PS electrospun fibres it is proved that by increasing the amount of TiO₂ nanoparticles, the glass transition temperature could be decreased.

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